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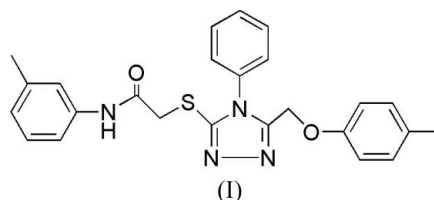
Key indicators

Single-crystal X-ray study
 $T = 294\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.043
 wR factor = 0.113
Data-to-parameter ratio = 15.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.3'-Methyl-2-[5-(4-methylphenoxy)methyl]-
4-phenyl-4*H*-1,2,4-triazol-5-ylsulfanyl]-
acetanilideIn the crystal structure of the title compound, $\text{C}_{25}\text{H}_{24}\text{N}_4\text{O}_2\text{S}$, intermolecular $\text{N}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds and $\text{C}-\text{H}\cdots\pi$ interactions link the molecules into a two-dimensional network parallel to the bc plane.

Received 16 October 2006

Accepted 29 October 2006

Comment

During the last few decades, considerable attention has been devoted to 1,2,4-triazole derivatives because of their comprehensive bioactivities, such as antimicrobial (Gulerman *et al.*, 2001), anti-inflammatory (Maxwell *et al.*, 1994), analgesic (Turan *et al.*, 1999), antitumor (Demirbas & Demirbas, 2002), antihypertensive (Paulvannan *et al.*, 2000), anticonvulsant (Husain *et al.*, 1987) and antiviral activities (Kritsanida *et al.*, 2002). The broad biological activities that the 1,2,4-triazoles show may be due to the presence of the $\text{N}=\text{C}-\text{S}$ unit (Omar *et al.*, 1986). In a continuation of our previous work on the syntheses and biological activities of 1,2,4-triazole derivatives (Liu *et al.*, 2006), we report here the synthesis and crystal structure of the title compound, (I) (Fig. 1).

The triazole ring, the C1–C6 benzene ring, the C12–C17 phenyl ring and the C19–C24 benzene ring are planar, with r.m.s. deviations of 0.006, 0.003, 0.004 and 0.008 Å, respectively. The C1–C6, C12–C17 and C19–C24 planes form dihedral angles of 29.7 (1), 79.2 (1) and 70.4 (1)°, respectively, with the N1–N3/C10/C11 plane.

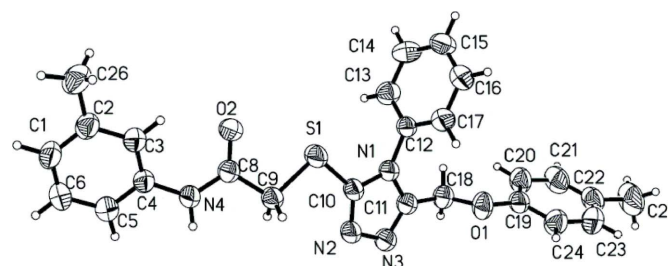
Two types of hydrogen bonds, *viz.* $\text{N}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{N}$, and $\text{C}-\text{H}\cdots\pi$ interactions involving the C1–C6

Figure 1

A plot of (I), with displacement ellipsoids drawn at the 50% probability level.

benzene ring are observed in the crystal structure (Table 1 and Fig. 2). The hydrogen bonds link the molecules into a two-dimensional network parallel to the *bc* plane (Fig. 3).

Experimental

N-(3-Methylphenyl)-2-chloroacetanilide (3 mmol) was added to a solution of 3-(4-methylphenoxy)methyl-4-phenyl-1,2,4-triazole (3 mmol) in ethanol (15 ml) and NaOH (3 mmol), and the mixture was refluxed for 1.5 h. After the mixture had been cooled to room temperature, the crude product was filtered and washed with distilled water (3×10 ml). Pure compound (I) was obtained by recrystallization from ethanol. Colourless single crystals were obtained by slow evaporation of an ethanol solution of (I) after about two weeks.

Crystal data

$C_{25}H_{24}N_4O_2S$	$Z = 4$
$M_r = 444.54$	$D_x = 1.275 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 16.8947 (18) \text{ \AA}$	$\mu = 0.17 \text{ mm}^{-1}$
$b = 8.2884 (9) \text{ \AA}$	$T = 294 (2) \text{ K}$
$c = 17.2401 (17) \text{ \AA}$	Block, colourless
$\beta = 106.382 (4)^\circ$	$0.30 \times 0.27 \times 0.17 \text{ mm}$
$V = 2316.1 (4) \text{ \AA}^3$	

Data collection

Bruker SMART CCD area-detector diffractometer	13217 measured reflections
φ and ω scans	4892 independent reflections
Absorption correction: multi-scan (SADABS, Bruker, 1998)	3186 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.951$, $T_{\max} = 0.972$	$R_{\text{int}} = 0.028$
	$\theta_{\text{max}} = 26.7^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.048P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.043$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.113$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.12$	$\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
4892 reflections	$\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$
320 parameters	Extinction correction: SHELXL97
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.0027 (8)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$ is the centroid of the C1–C6 benzene ring.

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N4-H4 \cdots N3^i$	0.89 (2)	2.09 (2)	2.982 (2)	176 (2)
$C14-H14 \cdots N2^ii$	0.93	2.62	3.329 (3)	134
$C17-H17 \cdots Cg1^{iii}$	0.93	2.72	3.508 (2)	143

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $-x + 1, -y, -z + 1$.

Atoms H1, H3–H6, H9A and H9B were located in a difference map and were refined isotropically [$N-H = 0.89 (2) \text{ \AA}$ and $C-H = 0.94 (2)–0.99 (2) \text{ \AA}$]. The remaining H atoms were positioned geometrically, with $C-H$ distances in the range $0.93–0.97 \text{ \AA}$, and refined using a riding model, with $U_{\text{iso}}(H) = 1.5U_{\text{eq}}(C)$ for methyl H atoms and $1.2U_{\text{eq}}(C)$ for the other H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Bruker, 1998); program(s) used to refine

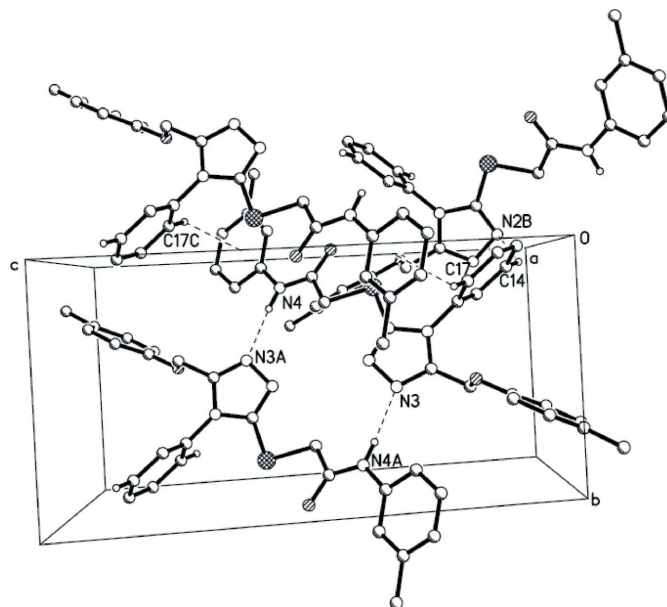


Figure 2

Part of the crystal structure of (I), showing $N-H \cdots N$, $C-H \cdots N$ and $C-H \cdots \pi$ interactions as dashed lines. Only the H atoms involved in the interactions are shown. Atoms labelled with the suffixes A, B and C are generated by the symmetry operations $(1-x, 1-y, 1-z)$, $(1-x, -\frac{1}{2}+y, \frac{1}{2}-z)$ and $(1-x, -y, 1-z)$, respectively.

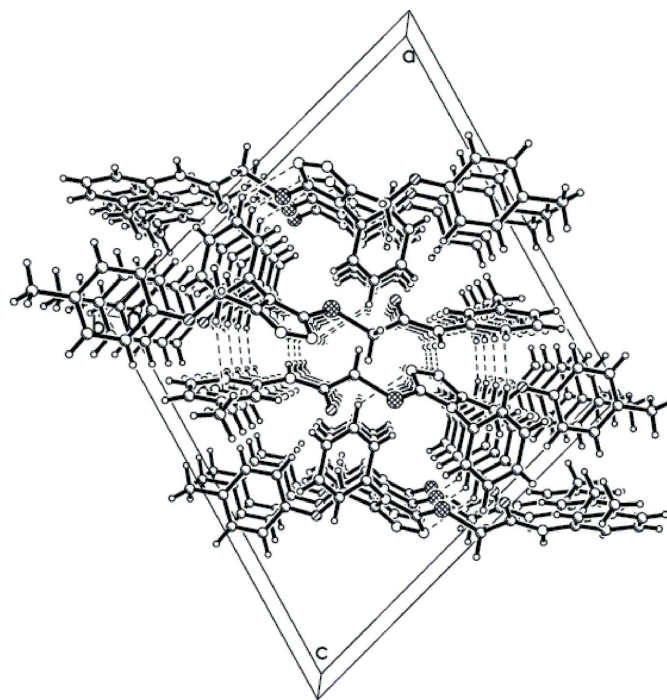


Figure 3

The crystal structure of (I). Hydrogen bonds are shown as dashed lines.

structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

The authors gratefully acknowledge the support of this work by the National Natural Science Foundation of China (grant Nos. 20371040 and 20671077), the Key Project of the Chinese Ministry of Education (grant No. 205161) and the

Foundation of Gansu Province (grant Nos. 3YS051-A25-010 and 3ZS061-A25-027).

References

- Bruker (1998). *SMART* (Version 5.0), *SAINT* (Version 4.0), *SADABS* (Version 2.0) and *SHELXTL* (Version 6.1). Bruker AXS Inc., Madison, Wisconsin, USA.
- Demirbas, N. & Demirbas, U. A. (2002). *Bioorg. Med. Chem.* **10**, 3717–3723.
- Gulerman, N. N., Dogan, H. N., Rollas, S., Johansson, C. & Elik, C. C. (2001). *Il Farmaco* **56**, 953–958.
- Husain, M. I., Amir, M. & Singh, E. (1987). *Indian J. Chem. Sect. B*, **26**, 251–254.
- Kritsanida, M., Mouroutsou, A., Marakos, P., Pouli, N., Papakonstantinou-Garoufalias, S., Pannecouque, C., Witvrouw, M. & Clercq, D. B. (2002). *Il Farmaco*, **57**, 253–257.
- Liu, H., Zhang, Y.-M., Lin, Q., Wei, T.-B. & Lu, T.-B. (2006). *Acta Cryst.* **E62**, o2890–o2892.
- Maxwell, J. R., Wasdahl, D. A. & Wolfson, A. C. (1994). *J. Med. Chem.* **27**, 1565–1570.
- Omar, A., Mohsen, M. E. & Wafa, O. M. (1986). *Heterocyclic Chem.* **23**, 1339–1341.
- Paulvannan, K., Chen, T. & Hale, R. (2000). *Tetrahedron*, **56**, 8071–8076.
- Turan, Z. G., Kaplancikli, Z. A., Erol, K. & Kilic, F. S. (1999). *Il Farmaco* **54**, 218–223.